

# Development of a New Route to 3-(Difluoromethyl)-1-methyl-1*H*-pyrazole-4-carboxylic Acid (DFPA) – A Key Building Block for Novel Fungicides

Poster #36 (Poster Award Winner), 22<sup>nd</sup> International Symposium on Fluorine Chemistry (ISFC-22), July 22–27, 2018, University of Oxford (UK)

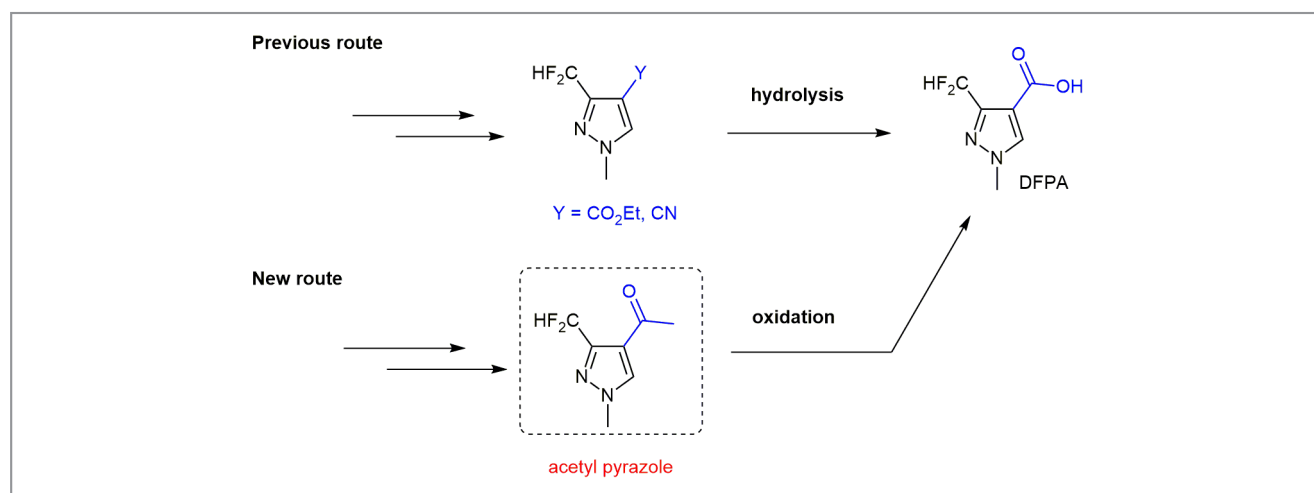
3-(Difluoromethyl)-1-methyl-1*H*-pyrazole-4-carboxylic acid (DFPA) is a key intermediate in the synthesis of succinate dehydrogenase inhibitors (SDHI), a new class of fungicides: six recently marketed compounds and four compounds currently in development have the same DFPA scaffold. Therefore, the global demand for this key intermediate is growing rapidly. For this reason, Yosuke Ochi and co-workers at the Asahi Glass Company (currently AGC Inc.) (Japan) started to develop an original route. Mr. Ochi said: “In most of the conventional routes (for a recent review, see *J. Fluorine Chem.* **2013**, *152*, 2–11), DFPA is synthesized by hydrolysis of a pyrazole ester (or nitrile), and there was no report on the preparation of DFPA from an acetyl pyrazole. Thus, we developed the new route using an acetyl pyrazole as a key intermediate of DFPA (Scheme 1).”

Mr. Ochi explained: “In our new route, each step of the reaction proceeds quantitatively and DFPA is obtained with very high purity. Moreover, we can utilize our in-house raw materials and technologies, such as difluoroacetyl fluoride (DFAF), chloroform, and NaOCl.” The starting material, dimethylaminovinyl methyl ketone (DMAB), is prepared via a well-known method using acetone, ethyl formate, and di-

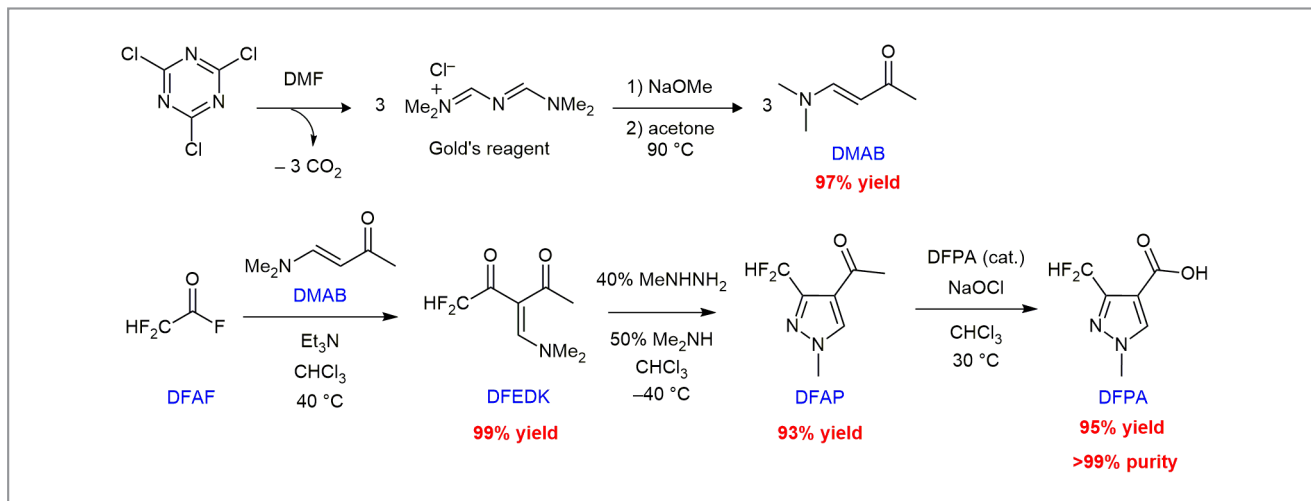
methylaniline. However, the group developed an alternative route to DMAB using Gold's reagent. “The new route is more cost-effective compared with the previous one (patent application WO2018074411),” remarked Mr. Ochi. He continued: “The difluoroacetylation of DMAB proceeds quantitatively using Et<sub>3</sub>N to trap HF. In the cyclization step, the addition of Me<sub>2</sub>NH controls the formation of several isomers and the trace amount of isomers generated can be removed efficiently by crystallization. The oxidation of an acetyl pyrazole with NaOCl is a very clean reaction, and DFPA is obtained in up to 99% purity. In this reaction, we use DFPA as a phase-transfer catalyst to make the reaction proceed smoothly (Scheme 2).”

This new method is practical and cost-effective for large-scale production. The group has already conducted pilot-scale production with this method.

Mr. Ochi concluded: “We would like to thank the R&D division in Chiba and Yokohama for their help in the process optimization and scale-up, and the analytical science team for identifying impurities. Finally, we are grateful to the manufacturing group in the Wakasa plant for their support during pilot-scale production.”



**Scheme 1** Approaches to the preparation of DFPA



Scheme 2 Preparation of DFPA using Gold's reagent

*Mattew Fank*

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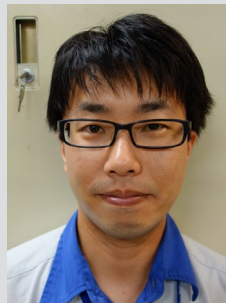
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