Mild and Versatile Nitrate-Promoted C-H Bond Fluorination

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■ Carbon-fluorine bond construction has consistently been of great interest to chemists due to the unique characteristics of fluorinated molecules and the synthetic challenges connected with the introduction of fluorine into organic molecules. Apart from the *ipso*-fluorination of pre-functionalized substrates, C-H bond fluorination is a prospective alternative and has received a great deal of attention in recent years. However, several challenges, for example, relatively harsh conditions, excess amounts of fluorinating agents (oxidants), narrow substrate scopes and poor selectivity still need to be addressed for this emerging area. Given the broad application of fluorinated

compounds in pharmaceuticals, agrochemicals and materials, a mild and versatile C–H bond fluorination protocol is highly desirable. Very recently, a novel and facile nitrate-promoted regioselective fluorination of aromatic and olefinic sp^2 -C–H bonds under mild conditions was described by Professors Dan-Qian Xu and Zhen-Yuan Xu, and Dr. Shao-Jie Lou from Zhejiang University of Technology (P. R. of China).

Generally, the fluorination reactions took place under very mild conditions (close to room temperature in most cases). Professor D.-Q. Xu said: "Conventionally, C–H bond fluorination requires harsh conditions with respect to the great



Scheme 1 Aromatic C-H bond fluorination



Scheme 2 Olefinic C-H bond fluorination

strength of both C–H and C–F bonds, so we were delighted to find that the current palladium-nitrate catalytic system enabled the process in a much milder manner."

"A catalytic amount of simple, non-toxic and cheap potassium nitrate served as a highly efficient promoter," said Professor D.-Q. Xu. "Actually, silver nitrate was found to be an efficient additive for this transformation at the outset of this program," she explained. Since silver salts were demonstrated to be incorporated in several C–F bond formation reactions, various silver salts were then screened. However, only silver nitrate and silver nitrite could successfully promote the reaction, whereas other silver salts were ineffective. Professor Z.-Y. Xu said: "The unique counter-anion effect led us to hypothesize that the nitrate anion might be the pivotal promoter. To our delight, nitrates were finally found to exhibit a unique ligand effect in this catalytic fluorination protocol." He continued: "Though the actual role of the nitrate is still not clear at this stage, we proposed that a highly active cationic Pd(NO₃)⁺ was generated in situ and initialized the C-H activation under mild conditions. Meanwhile, the poorly nucleophilic nitrate additives might also be responsible for the selective reductive elimination of the C-F bond from the Ar-Pd(IV)-F intermediate (*Angew. Chem. Int. Ed.* **2011**, *50*, 1478)."

The process shows a remarkably broad substrate scope for both aromatic and olefinic sp²-C–H bonds. In general, both electron-donating and electron-withdrawing functional groups were well tolerated by cautiously adjusting the reaction temperature of the aromatic C–H bond fluorination. Moreover, good mono-/di-fluorination selectivity could also be achieved by controlling the reaction temperature. Professor D.-Q. Xu said: "Notably, the first example of chelation-assisted olefinic sp²-C–H bond fluorination was also reported in this paper. Various functionalities, for example, alkyl-, halo-, and aryl-substituted α , β -unsaturated oximes were well tolerated and furnished the β -fluorinated products in good yields at room temperature.



Scheme 3 C-H fluorination of ketones

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In addition, the oximyl directing group can be efficiently installed or removed, which would provide opportunities for further derivatization of the fluorinated ketone products. Gram-scale C–H bond fluorination proceeded smoothly with a reduced loading of fluorinating agent.

Professor Z.-Y. Xu said: "Given the mild conditions and universality of this method, the present fluorination protocol may enable the late-stage fluorination of more complex substrates without touching the other functional groups. Attempts to apply this system to more substrates are ongoing in our lab.

In conclusion, we have developed a novel nitrate-promoted fluorination system, which features broad substrate scope, good functional group tolerance and simple operations," said Professor D.-Q. Xu. "It should pave the way for mild and versatile C–H bond fluorination in synthetic and pharmaceutical chemistry."



About the authors



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Prof. D.-Q. Xu

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