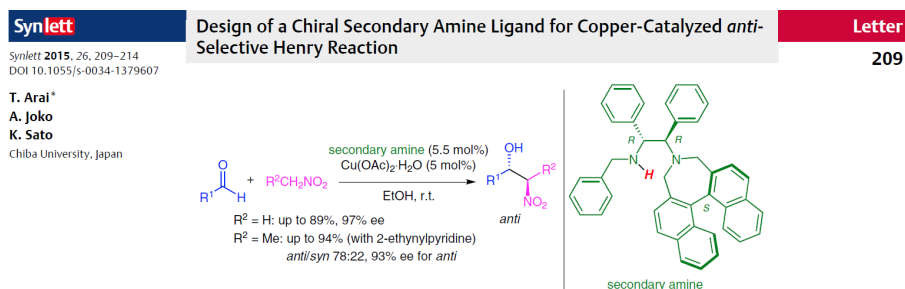
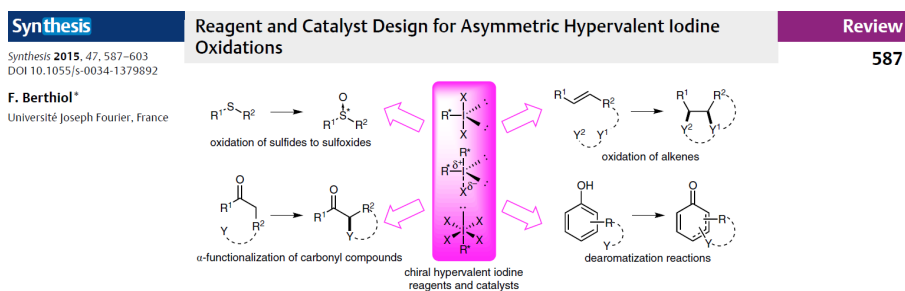
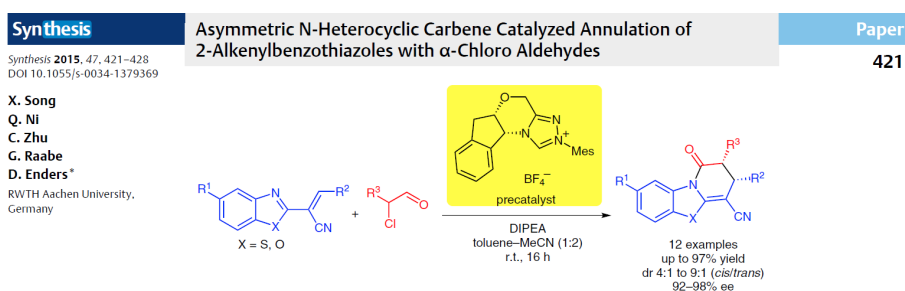


## Graphical Abstracts

The graphical abstract is the most important scheme of your manuscript and determines whether a reader continues to read on about your research. A well-prepared graphical abstract is the best advertisement for your paper and draws attention to your findings. Synthetic chemistry is visual, and most of us scan the literature using just abstracts. Time spent on a nice graphical abstract is time well spent.

The take-home message of your paper should be clear from the graphical abstract. Please use colors to clearly highlight substantive information and to make it easy to interpret. Think about what's interesting to the reader. Yield ranges, the number of examples reported, the reaction conditions used, and an indication of the substrate scope. It should be possible for readers to identify the relevance of the research described in your paper to their own interest. The inclusion of color is free of charge of course, as is the publication of your graphical abstract on the journal cover, if selected! All electronic versions will be in color. You may notice, however, than in the print version, the graphical abstract within the paper may appear in black and white.

The following ten samples should help you to prepare your own graphical abstract. The maximum dimensions are 11 × 5 cm (4.3 × 2.0 in.). We like the first example because the tasteful use of color helps to illustrate the chemistry. It is immediately clear that the enantioselectivity is excellent, that the yields are very good, and that the substrate scope is reasonable. In addition, it is evident that the reaction conditions are straightforward, and that the reaction should be operationally simple.



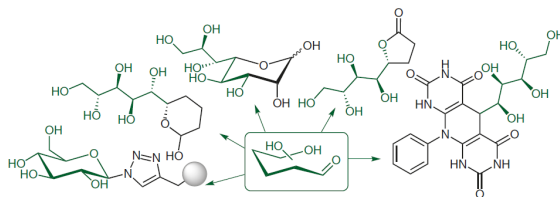
**Synlett**

 Synlett 2015, 26, 421–425  
 DOI 10.1055/s-0034-1379979

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**Unprotected Carbohydrates as Starting Material in Chemical Synthesis:  
 Not Just a Challenge but an Opportunity**
**Synfacts**

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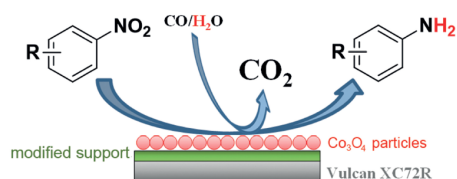

**Synlett**

 Synlett 2015, 26, 313–317  
 DOI 10.1055/s-0034-1380003

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**Reduction of Nitroarenes Using CO and H<sub>2</sub>O in the Presence of a Nano-  
 structured Cobalt Oxide/Nitrogen-Doped Graphene (NGr) Catalyst**
**Cluster**

313

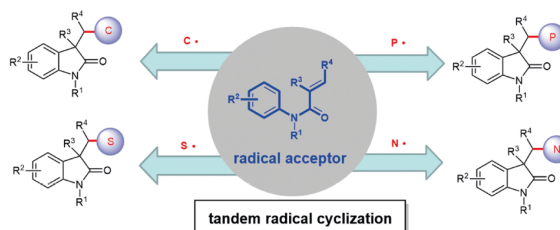

**Synthesis**

 Synthesis 2015, 47, 604–629  
 DOI 10.1055/s-0034-1378944

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**Tandem Radical Cyclization of *N*-Arylacrylamides: An Emerging  
 Platform for the Construction of 3,3-Disubstituted Oxindoles**
**Short Review**

604

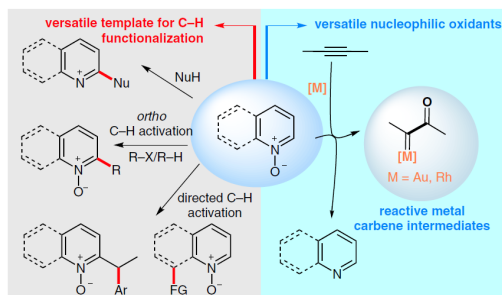

**Synthesis**

 Synthesis 2015, 47, 289–305  
 DOI 10.1055/s-0034-1379884

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**Recent Developments in the Chemistry of Heteroaromatic *N*-Oxides**
**Review**

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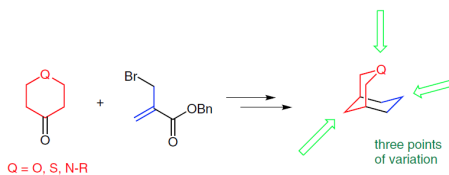

**Synthesis**

 Synthesis 2015, 47, 367–376  
 DOI 10.1055/s-0034-1379456

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**An Expedient and Practical Approach to Functionalized 3-Aza-, 3-Oxa-,  
 and 3-Thiabicyclo[3.3.1]nonane Systems**
**Paper**

367



**Synthesis**

## A Scalable, Chromatography-Free Synthesis of Benzotetramisole

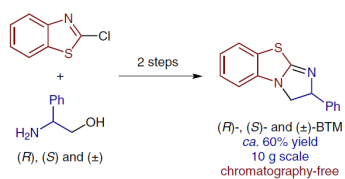
PSP

*Synthesis* **2015**, 47, 34–41  
DOI 10.1055/s-0034-1378931

34

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**Synthesis**

## Effect of Substituents and Stability of Transient Aluminum–Aminals in the Presence of Nucleophiles

Feature

*Synthesis* **2015**, 47, 175–180  
DOI 10.1055/s-0034-1379635

175

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